Docket No.: 07206-1180 Client Ref. No.: 99E142

CLAIMS

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1. A method of making aluminum oxynitride, the method comprising: introducing aluminum oxide particles into a chamber; dispersing the particles within the chamber; and

forming the aluminum oxynitride comprising passing nitrogen gas over the dispersed particles.

- 2. The method of claim 1, wherein forming the aluminum oxynitride comprises heating the particles.
 - 3. The method of claim 1, further comprising introducing carbon into the chamber to form a mixture comprising aluminum oxide and carbon.
 - 4. The method of claim 1, further comprising introducing a reducing agent into the chamber to form a mixture comprising aluminum oxide and the reducing agent.
 - 5. The method of claim 1 wherein forming the aluminum oxynitride comprises heating the mixture.
 - 6. A method of making aluminum oxynitride, the method comprising: introducing a mixture comprising aluminum oxide and carbon into a chamber; agitating the mixture within the chamber; and heating the mixture to make aluminum oxynitride.
 - 7. The method of claim 6, further comprising: introducing nitrogen gas into the chamber.
- 1 8. The method of claim 6, wherein agitating the mixture comprises rotating the chamber.

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9. The method of claim 6, further comprising: cooling the aluminum oxynitride;

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- 10. The method of claim 6, further comprising:
- 2 forming the aluminum oxynitride into a transparent structure.
- 1 11. The method of claim 10, wherein forming the aluminum oxynitride comprises:
- 2 forming a green body comprising the aluminum oxynitride; and
- 3 sintering the green body.
 - 12. The method of claim 11, further comprising:
 - isostatically pressing the sintered green body under heat.
 - 13. The method of claim 6, wherein the aluminum oxynitride comprises Al₂₃.
 - $_{1/3x}O_{27+x}N_{5-x}$, where $0.429 \le x \le 2$.
 - 14. A method of making aluminum oxynitride, the method comprising:

introducing a first reaction mixture comprising aluminum oxide and carbon into a chamber;

agitating the first reaction mixture within the chamber;

heating the chamber to a temperature to form aluminum oxynitride from the first reaction mixture;

removing the aluminum oxynitride while maintaining the temperature of the chamber;

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introducing a second reaction mixture comprising aluminum oxide and carbon into

the chamber while maintaining the temperature of the chamber.

- 15. The method of claim 14, further comprising:
- 2 introducing nitrogen gas into the chamber.

- 16. The method of claim 14, wherein introducing the first reaction mixture comprises introducing the first reaction mixture from a hopper.
- 17. The method of claim 14, wherein agitating the first reaction mixture comprises 1 rotating the chamber. 2
- 18. The method of claim 14, wherein the chamber comprises an exit opening and 1 removing the aluminum oxynitride comprises retracting a plunger within the chamber, 2 3 thereby allowing the aluminum oxynitride to flow through the exit opening.
- 19. The method of claim 14, further comprising: 1 2 forming the aluminum oxynitride into a transparent structure.
 - 20. The method of claim 19, wherein forming the aluminum oxynitride comprises: forming a green body comprising the aluminum oxynitride; and sintering the green body.

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- 21. The method of claim 20, wherein forming the aluminum oxynitride comprises: isostatically pressing the sintered green body under heat.
- 22. The method of claim 14, wherein the aluminum oxynitride comprises Al₂₃. 1
- $_{1/3x}O_{27+x}N_{5-x}$, where $0.429 \le x \le 2$. 2
 - 23. An aluminum oxynitride made according to the method of claim 6.
- 24. The aluminum oxymitride of claim 23, wherein the aluminum oxymitride 1 comprises $Al_{23-1/3x}O_{27+x}N_{5,x}$, where $0.429 \le x \le 2$. 2
 - 25. A method of making aluminum oxynitride, the method comprising:

heating a chamber;

continuously introducing a reaction mixture comprising aluminum oxide and carbon

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into the chamber;

agitating the reaction mixture within the chamber; and